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Note Taking Guide: Episode 303 Name Aditi Acharya Bohr's Energy Levels

- Electrons in certain energy levels
- Lower energy levels: closer to nucleus
- Higher energy levels: further from nucleus
- Ground State: electron in lowest energy level possible
- Excited Atom
- Atom has absorbed energy
- Excited state is unstable
- Atom soon emits same amount of energy absorbed
- energy seen as visible light

Wave Description of Light Wavelength (?): distance between corresponding points ...

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- Electrons in certain energy levels
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- Ground State: electron in lowest energy level possible
- Excited Atom
- Atom has absorbed

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